

# Experimental and Numerical Evaluation of Pressurized, Lean Hydrogen-Air Flame Stability with Carbon Dioxide Diluent

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## **ABSTRACT**

The potential to combine steam-methane reforming for large-scale hydrogen production with CCS provides two unique vectors ( $H_2$  and  $CO_2$ ) for utilization in decarbonized gas turbine power generation. Therefore, a study has been conducted to evaluate the use of  $CO_2$  as an additive in lean premixed (LPM)  $H_2$ -air swirl flames. Chemical kinetics modelling indicates that the use of  $CO_2$  in the premixed reactants reduces  $H_2$ -air laminar flame speed ( $S_L$ ) and adiabatic flame temperature (AFT) within the well-characterized range of preheated LPM  $CH_4$ -air flames, albeit in markedly different proportions; nearly 70% vol  $CO_2$  in the fuel is required for a reduction in  $S_L$  to equivalent  $CH_4$ -air values, while approximately 30% vol  $CO_2$  in the fuel is required for an equivalent reduction in AFT, impacted by the increased heat capacity of  $CO_2$ . A generic gas turbine swirl burner is therefore utilized to experimentally investigate the influence of  $CO_2$  dilution on pressurized (25 kW/bar), preheated (573 K), LPM  $H_2$ -air flame stability using high-speed OH\* chemiluminescence and dynamic pressure sensing. The influences of extinction strain rate and Lewis number are suggested to characterize, both experimentally and numerically, the observed lean flame behavior, in particular as extinction strain rate has been shown to be non-monotonic with pressure for highly-reactive and diffuse fuels such as  $H_2$ .

#### **INTRODUCTION**

The use of hydrogen in lean premixed gas turbine power generation presents a number of interesting challenges for the use of established burner technologies. An experimental and numerical study is therefore conducted in a **fully premixed**, **preheated** (**T2 = 573 K ± 5 K**), **pressurized** (**25 kW/bar**) generic swirl burner to investigate the use of CO<sub>2</sub> as a premixed fuel additive, given its ability to moderate the combustion process; for example, as shown in **Figure 1**, which plots modelled laminar flame speed,  $S_L$ , as a function of fuel CO<sub>2</sub> (%vol) and equivalence ratio ( $\varphi$ ) with the equivalent CH<sub>4</sub>-air flame speed at  $\varphi$  = 0.6.

In addition to the increase in reactivity observed in Figure 1,  $H_2$ -air flames also exhibit enhanced thermodiffusive effects, making them less resistant to extinction under highly stretched conditions. This effect, however, has been shown experimentally to be non-monotonic with pressure, as shown in **Figure 2**, by Niemann et al. [3,4].

### **RESULTS**

**Figure 5**: The use of H<sub>2</sub> reduces the lean blowoff (LBO) equivalence ratio and requires stabilization at higher *Re*. Up to 40% vol CO<sub>2</sub> in the premixed reactants at  $\varphi > 0.70$  is achieved.

**Figure 6**: For increasing  $\varphi$ , the achievable CO<sub>2</sub> dilution increases, but the stable operational range reduces. At  $\varphi < 0.3$ , pure H<sub>2</sub>-air flames were stabilized, however burner exhaust temperatures were insufficient (<1100 K), suggesting a future increase in 25 kW/bar scaling.

**Figure 7**: With increasing pressure, the  $CO_2$  dilution level at LBO is nonmonotonic, which suggests a relationship with the extinction strain rate (see **Figure 2**) due to the high turbulence.





Figure 1: Laminar flame speed ( $S_L$ ) modeling of CO<sub>2</sub> dilution effects on H<sub>2</sub>-air flame and comparison with equivalent CH<sub>4</sub>-air flame at  $\varphi$  = 0.6 (red line), using Li et al. [1] and GRI-Mech 3.0 [2] mechanisms, respectively

Figure 2: Experimental extinction strain rate measurements for CH<sub>4</sub>-N<sub>2</sub>-air and H<sub>2</sub>-N<sub>2</sub>-air counterflow diffusion flames at elevated pressure [3,4]

### **EXPERIMENTAL METHODS**

A comparison of burner operability between CH<sub>4</sub>-air and H<sub>2</sub>-CO<sub>2</sub>-air flames is first conducted at near-ambient burner inlet pressure (P2 = 1.1 bara) with flame stabilization mechanisms and flame instabilities identified using high-speed OH\* chemiluminescence (4 kHz, 10 µs gate, 4.65 pix/mm) alongside in-burner dynamic pressure measurement (4 kHz). Then, a single lean equivalence ratio ( $\varphi$  = 0.25) is selected to investigate the influence of P2 on lean H<sub>2</sub>-CO<sub>2</sub>-air flame stability and stretch behavior. Refer to **Table 1** for experimental operating conditions.

Table 1: Experimental o	perating conditions	for fully premixed	CH₄-air and H	<sub>2</sub> -CO <sub>2</sub> -air flames
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<b>Premixed Reactants</b>	ṁ <sub>air</sub> (g/s)	ṁ <sub>fuel</sub> (g/s)	т <sub>со2</sub> (g/s)	T2 (± 5 K)	P2 (bara)	P <sub>therm</sub> (kW)	φ	Re (x10 <sup>3</sup> )
CH₄-air	10.8 - 16.4	0.5	0	573	1.1	25	0.53 - 0.80	12 - 18
H <sub>2</sub> -CO <sub>2</sub> -air	100-715	02-052	0 - 14 37	573	11-275	25 - 62	0 25 - 0 72	27 - 90

**Figures 8 and 9**: Flame stabilization prior to LBO is characterized by flame attachment (8.a and 8.d) and detachment (8.b and 8.c) from the burner exit nozzle, with significantly reduced OH\* intensity in the H<sub>2</sub>-CO<sub>2</sub>-air case. CH<sub>4</sub> heat release (9.a) and dynamic pressure (9.b) fluctuation near LBO have similar peak frequencies, with heat release fluctuation damped in H<sub>2</sub>-CO<sub>2</sub> flames.



Figure 8: Instantaneous OH\* chemiluminescence images near LBO for  $CH_4$ -air (a,b) and  $H_2$ - $CO_2$ -air (c,d) flames. Note burner exit nozzle in grey, flow from left to right

### **CONCLUSIONS AND FUTURE WORK**

Figure 9: Power spectral density of OH\* chemiluminescence fluctuation (a) and dynamic pressure fluctuation (b) at near LBO conditions from Figure 8

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All experimental work has been conducted utilizing the 2<sup>nd</sup> Generation High-Pressure Generic Swirl Burner (HPGSB-2) installed within the High-Pressure Optical Chamber (HPOC), refer to **Figure 3** and **Figure 4**, respectively.



Figure 3: Schematic of HPGSB-2 assembly with critical features identified Figure 4: Photograph of HPOC with HPGSB-2 and high-speed OH\* chemiluminescence system (Phantom v1212 camera, SIL-3 image intensifier, 78 mm UV lens and 310 nm bandpass filter) Fully premixed, lean H<sub>2</sub>-air flames were successfully stabilized in a swirl burner with sufficient levels of CO<sub>2</sub> dilution (up to 75% vol fuel / 40% vol premix) and turbulence ( $Re_{H2} \approx 2^*Re_{CH4}$ ). Fully premixed, pure H<sub>2</sub>-air flames were successfully stabilized at low equivalence ratios ( $\varphi < 0.3$ ) with sufficiently high burner exit nozzle velocity.

Flame attachment and detachment from the burner exit nozzle precedes lean blowoff, with heat release fluctuations and pressure fluctuations near LBO dampened by the addition of  $CO_2$  in the premixed reactants.

Maximum levels of  $CO_2$  dilution in H<sub>2</sub>-air flames near LBO exhibit a non-monotonic influence of pressure, suggesting that thermodiffusive effects dominate under these highly stretched conditions, in agreement with the work of Niemann et al. [3,4].

Future work to include operation with higher  $CO_2$  mass flow at higher  $\varphi$  to achieve representative GT burner exit temperatures, operation at higher pressures to further evaluate stretch effects, an evaluation of  $NO_x/CO$  emissions, and the development of a pressurized counterflow burner system for fundamental study of H<sub>2</sub>-CO<sub>2</sub>-air extinction behavior.

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